Name	date
A) catabolic pathways.	se stored energy by breaking down complex molecules are known as
B) anabolic pathways.	
C) bioenergetic pathways.	
D) endergonic pathways.	
2 In an oxidation-reduction reac	tion, the reducing agent
A) gains electrons and gains potential	
B) gains electrons and loses potential	= -
C) loses electrons and loses potential e	energy.
D) loses electrons and gains potential	
3 As a result of an oxidation-red	fuction reaction the oxidizing agent
A) gains electrons and gains potential	
B) gains electrons and loses potential of	
C) loses electrons and loses potential e	
D) loses electrons and gains potential	
4 As a result of the transfer of an	n electron from a less electronegative atom to a more electronegative
atom,	
A) the more electronegative atom is re	
B) the more electronegative atom is re	
C) the more electronegative atom is or	
D) the more electronegative atom is or	kidized, and energy is released.
5 The complete reactions of cell	ular respiration in the presence of oxygen ($C_6H_{12}O_6 + 6 O_2 \rightarrow 6 CO_2 + 6 O_3 \rightarrow 6 O_3 + $
6 H ₂ O + energy) result in which of the	e following?
A) oxidation of O ₂ and reduction of H	I ₂ O
B) oxidation of $C_6H_{12}O_6$ and reduction	on of O ₂
C) reduction of CO_2 and oxidation of	O_2
D) reduction of $C_6H_{12}O_6$ and oxidati	on of CO ₂
6 Which of the following statem	nents about NAD+ is true?
A) NAD+ is the source of electrons us	
	· · ·
B) NAD+ has more chemical energy to	
C) NAD+ is oxidized by the action of	
D) NAD+ is reduced to NADH during	glycolysis.
7 In animal cells, glycolysis occ	eurs in the
A) cytosol.	
B) outer mitochondrial membrane.	
C) inner mitochondrial membrane.	
D) mitochondrial matrix.	
E) nucleus.	

 8 The ATP produced in glycolysis is generated by A) chemiosmosis. B) electron transport. C) photophosphorylation. D) oxidative phosphorylation. E) substrate-level phosphorylation.
 9 The ATP produced in the citric acid cycle is generated by A) chemiosmosis. B) electron transport. C) photophosphorylation. D) oxidative phosphorylation. E) substrate-level phosphorylation.
 10 The oxygen consumed during cellular respiration is involved directly in which process or event? A) glycolysis B) accepting electrons at the end of the electron transport chain C) the citric acid cycle D) the oxidation of pyruvate to acetyl CoA E) the phosphorylation of ADP to form ATP
11 The complete oxidation of glucose in aerobic respiration occurs through which of the following sequence of metabolic reactions? A) glucose → citric acid cycle → glycolysis → pyruvate oxidation → electron transport chain B) glucose → pyruvate oxidation → glycolysis → electron transport chain → citric acid cycle C) glucose → glycolysis → pyruvate oxidation → citric acid cycle → electron transport chain D) glucose → glycolysis → citric acid cycle → pyruvate oxidation → electron transport chain E) glucose → pyruvate oxidation → citric acid cycle → glycolysis → electron transport chain
 12 During glycolysis, when each molecule of glucose is catabolized to two molecules of pyruvate, most of the potential energy contained in glucose is A) transferred to ADP, forming ATP. B) released as heat. C) retained in the two pyruvates. D) stored in the NADH produced.
13 Starting with one molecule of glucose, the energy-containing products of glycolysis are A) 2 NAD+, 2 pyruvate, and 2 ATP. B) 2 NADH, 2 pyruvate, and 2 ATP. C) 2 FADH2, 2 pyruvate, and 4 ATP. D) 6 CO2, 2 pyruvate, and 2 ATP. E) 6 CO2, 2 pyruvate, and 30 ATP.
 14 In the complete reactions of aerobic respiration, the energy for the majority of ATP synthesis is provided by A) transfer of electrons from organic molecules to acetyl CoA. B) high-energy phosphate bonds from organic molecule intermediates in the citric acid cycle. C) splitting water to produce oxygen. D) a proton gradient across the mitochondrial inner membrane. E) the production of carbon dioxide and oxygen in the electron transport chain.

15 What is the source of the oxygen used to form water in the complete reactions of cellular respiration? A) carbon dioxide (CO ₂)
B) glucose (C ₆ H ₁₂ O ₆)
C) pyruvate (C ₃ H ₃ O ₃ -)
D) molecular oxygen (O ₂)
16 In chemiosmosis, what is the most direct source of energy that is used to convert ADP $+$ \textcircled{P}_i to ATP? A) energy released as electrons flow through the electron transport system B) energy released from substrate-level phosphorylation C) energy released from dehydration synthesis reactions D) energy released from movement of protons down their electrochemical gradient through ATP synthase
17 In liver cells, the inner mitochondrial membranes are about five times the area of the outer mitochondrial membranes. What purpose must this serve?A) It increases the surface area for glycolysis.B) It increases the surface area for the citric acid cycle.C) It increases the surface area for oxidative phosphorylation.
D) It increases the surface area for substrate-level phosphorylation.
18 Which of the following occur(s) in the cytosol of a eukaryotic cell? A) glycolysis and fermentation B) fermentation and chemiosmosis C) oxidation of pyruvate to acetyl CoA D) citric acid cycle E) oxidative phosphorylation
19 Which of the following occur(s) in mitochondria? A) glycolysis and fermentation B) fermentation and chemiosmosis C) glycolysis and oxidation of pyruvate to acetyl CoA D) oxidation of pyruvate to acetyl CoA and the citric acid cycle E) fermentation and oxidative phosphorylation
20 Which metabolic pathway is common to both cellular respiration and fermentation? A) the oxidation of pyruvate to acetyl CoA B) the citric acid cycle C) oxidative phosphorylation D) glycolysis E) chemiosmosis
21 Yeast cells grown anaerobically can obtain energy by fermentation, which results in the production of A) ATP, NADH, and pyruvate. B) ATP and lactate. C) ATP, CO ₂ , and lactate. D) ATP, CO ₂ , and ethanol. E) ATP, CO ₂ , and acetyl CoA.

- 22. ____ Why is glycolysis considered to be one of the first metabolic pathways to have evolved?
- A) It produces much less ATP than does oxidative phosphorylation.
- B) It does not involve organelles or specialized structures, does not require oxygen, and is present in most organisms.
- C) It is found in prokaryotic cells but not in eukaryotic cells.
- D) It relies on chemiosmosis, which is a metabolic mechanism present only in prokaryotic cells.
- 23. ____ A mutation in yeast makes it unable to convert pyruvate to ethanol. How will this mutation affect these yeast cells?
- A) The mutant yeast will produce lactate under anaerobic conditions.
- B) The mutant yeast will be unable to grow anaerobically.
- C) The mutant yeast will be unable to grow aerobically.
- D) The mutant yeast will grow anaerobically only when provided glucose.
- E) The mutant yeast will be unable to metabolize glucose.
- 24. ____ During intense exercise, as skeletal muscle cells switch to fermentation, the human body will increase its catabolism of
- A) fats only.
- B) carbohydrates only.
- C) proteins only.
- D) fats, carbohydrates, and proteins.
- E) fats and proteins only.

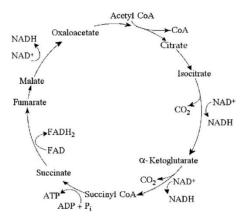


Figure 7.1

- 25. ____ Starting with one molecule of isocitrate and ending with fumarate, how many ATP molecules can be made through substrate-level phosphorylation (see Figure 7.1)?
- A) 1
- B) 2
- C) 11
- D) 12
- 26. ____ For each mole of glucose (C₆H₁₂O₆) completely oxidized by cellular respiration, how many moles of CO₂ are released in the citric acid cycle (see Figure 7.1)?
- A) 2
- B) 3
- C) 4
- D) 6
- E) 12

27. ____ For each molecule of glucose that is metabolized by glycolysis and the citric acid cycle (see Figure 7.1), what is the total number of NADH + FADH2 molecules produced?

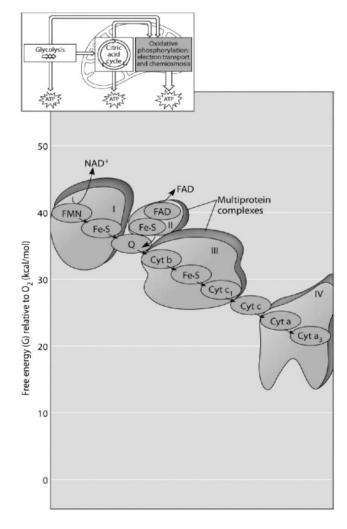
A) 4

B) 5

C) 6

D) 10

E) 12



- 28. ____ Figure 7.2 shows the electron transport chain. Which of the following is initially added to the chain with the highest free energy?
- A) oxygen
- B) FADH₂
- C) NADH
- D) CO₂
- E) water
- 29. ____ Which of the following is an accurate description of the events that occur along the electron transport chain depicted in Figure 7.2?
- A) ATP is generated directly at three points in the pathway.
- B) Electron transfer is directly coupled to chemiosmosis.
- C) Each electron transfer between carriers results in oxidation of one carrier and reduction of another.
- D) The potential energy of electrons increases at each step in the pathway.

30 O2?	Which of the protein complexes labeled with Roman numerals in Figure 7.2 will transfer electrons to
B) co C) co	omplex I omplex II omplex III omplex IV
	ll of the complexes can transfer electrons to O2.
31.	
to th	ording to the chemiosmotic model proposed by Peter Mitchell in 1961, an electrochemical gradient is linked be synthesis of ATP in mitochondria. Construct an explanation of the chemiosmotic model by doing each of following.
(a)	Make a claim about the role of the inner mitochondrial membrane in ATP synthesis.
(b)	Present ONE piece of evidence that supports the role you proposed in part (a).
	Provide reasoning to explain how the evidence you presented in part (b) supports the claim you made in part (a).